

THE INFLUENCE OF NANOPARTICLES ON THE DIELECTRIC DISSIPATION FACTOR AND LIGHTNING PROPERTIES IN PALM OIL-BASED NANOFLUIDS

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Abstract

In this study, the influence of nanoparticles when mixed with insulating palm oil is investigated in terms of $\tan \delta$ or the dielectric dissipation factor and lightning properties. Three different nanofluid sample types are derived from the dispersion of zinc oxide (ZnO), titanium dioxide (TiO₂) and barium titanate (BaTiO₃). The nanoparticle concentrations tested were 0.01% and 0.03% while the base fluid was unmodified palm oil. Evaluation of $\tan \delta$ was performed in accordance with the IEC 60247 standard via the use of the $\tan \delta$ meter (model SOKEN: DAC-IM-D6) and the lightning impulse breakdown voltage were based on the IEC 60897 standard. In order to test the electrical insulation qualities of the palm oil over a range of temperatures from 35-90°C for each of the $\tan \delta$ values revealed the dielectric loss tangent, and the findings lightning properties was performed at the room temperature. Significant increases of the palm oil with nanoparticles in comparison to unmodified palm oil. These results suggested that further investigation would be worthwhile to better understand the effects of the nanoparticles in palm oil upon $\tan \delta$ and the lightning impulse breakdown voltage characteristics in order to make additional improvements.

Keywords: Dielectric dissipation factor, Lightning impulse voltage, Nanofluids, Palm oil, Temperature.

1. Introduction

Mineral oils are the most important of the various liquid dielectrics which are applied in power equipment. These oils are typically refined mixtures of various hydrocarbons which can be produced via the fractional distillation of petroleum. They have properties which vary according to the chemical composition. One type of liquid insulating oil based on petroleum is transformer oil, which is refined precisely to address the need for both dielectric and the transfer of heat when used in transformers. Reliability and efficiency are vital for electrical transmission and distribution systems and the transformers used must have a working lifespan exceeding forty years. The type of oil used will be critical in determining the ability of a transformer to work effectively throughout that period of time. Over the past hundred years, the most frequently employed electrical insulating liquids have been petroleum-based transformer oils. The oil quantity required by transformers lies within quite a wide range, related primarily to the transformer power rating [1].

As a consequence of environmental concerns, recent studies have sought to find alternatives to liquid insulating materials, with insulating oils offering certain more environmentally friendly qualities. Among the most common is natural oil, which is both biodegradable and suitable for use in transformers. It is an oil based on seeds and contains fatty acid natural oil triglycerides [2]. The natural oil has been used as the dielectric materials in transformers on a small or medium scale where the voltage does not exceed 66 kV. In recent years have been conducted the studies to investigate the properties of natural oil [3], examining the ageing, safety, and environmental effects of its use, as well as the electrical performance of vegetable oil. It has been reported that transformers filled with vegetable oil can be successfully used at various voltage levels [4]. Ageing in cellulose insulation can potentially be mitigated by the use of vegetable oil, as the natural oil will provide water scavenging activity and hydrolytic protection. In addition to natural ester, it is suggested that palm oil is another form of vegetable oil which could be used in transformers as the insulating materials.

Palm oil has chemical attributes which are similar to those of natural ester. It is readily biodegradable, offers flash points and high fire and non-toxic. These qualities can be attributed to the fact that palm-based oils are of a food grade type [4]. Many research studies have been performed lately to examine the use of nanotechnology to enhance the properties of dielectric insulating fluids used in transformers. Oils have been combined with various kinds of nanoparticles in order to enhance the heat transfer capabilities, as well as the dielectric and thermophysical properties [5].

In this study, experiments were carried out to examine the dielectric dissipation factor, known as $\tan \delta$ and lightning properties, for widely used palm oil containing three different kinds of nanoparticles: zinc oxide (ZnO), titanium dioxide (TiO₂), and barium titanate (BaTiO₃). The $\tan \delta$ and the lightning impulse characteristics for the palm oil samples with nanoparticles added were tested and measured at a range of temperatures from 35 °C to 90 °C for $\tan \delta$ and the lightning experiments was performed at the room temperature. The study sought to test the notion that certain quantities of certain nanoparticles might increase the dielectric qualities of palm oil.

2. Methods and Materials

2.1. Tan δ (dielectric dissipation factor)

Tan δ , also known as the dielectric dissipation factor or the loss factor, is the parameter which describes the dielectric qualities of the oil. The level of them of the palm oil dissipation factor at its power frequency matches that of palm oil resistivity, and the same factors and testing conditions influence these two measures [6]. Furthermore, the tan δ is indicative of total dielectric power loss, and thus it serves as a useful estimator of capacitor power losses. Tan δ also acts as a function of the temperature of a dielectric, as well as the magnitude and frequency of the applied voltage since there is an effect on both conductivity (κ) and processes of polarization in dielectrics [1].

A single capacitor can represent the transformer insulation, whereby for dielectrics at AC voltage, the current (I) of the voltage (V) will lead by an angle of around $\varphi \approx 90^\circ$. As a consequence of conductivity losses and polarization losses, there can be a deviation in the phase angle φ from 90° by the loss angle (δ). Accordingly, the application of alternating voltage to the insulation of a distribution transformer involving insulating palm oil leads to the appearance of a current, comprising resistive (I_δ) and capacitive (I_C) elements, as can be observed in Fig. 1 [7, 8].

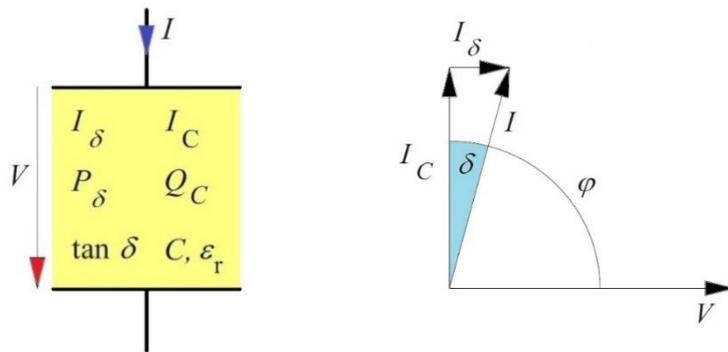


Fig. 1. Dissipation factor phasor diagrams, employing complex AC calculation techniques.

The loss factor or dissipation factor can serve to define the loss angle δ , as indicated by Eqs. (1) to (4).

$$\tan \delta = \frac{I_\delta}{I_C} \quad (1)$$

Using power quantities

$$P_\delta = VI_\delta \quad (2)$$

and

$$Q_C = VI_C \quad (3)$$

While the loss or dissipation factor can be expressed as

$$\tan \delta = \frac{P_\delta}{Q_C} \tag{4}$$

Furthermore, to establish the dissipation factor for the transformer insulation, the resistance and capacitance must be taken into consideration, and can be represented as parallel or series equivalent circuits, where V is applied alternating voltage and I is a current that will flow in circuits, as presented in Fig. 2 [7, 8].

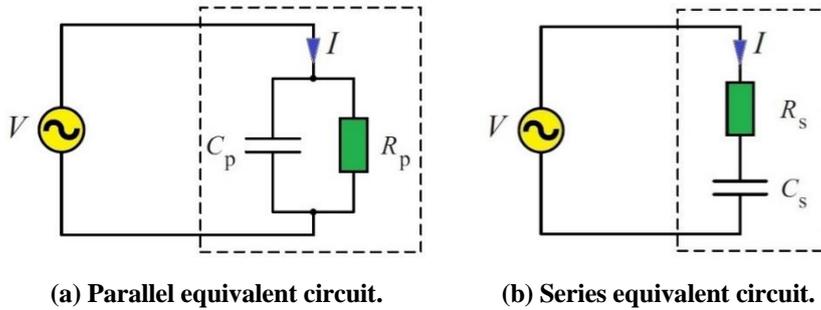


Fig. 2. Single equivalent capacitance and resistance.

From Eq. (1), the dissipation factors are derived from the ratio of power converted to resistance to reactive power allocated to capacitance [8]. In the case of the parallel equivalent circuit, the calculation considers the small $\tan \delta$ values such that the resulting $\tan \delta$ will be approximately equal, as shown in Eq. (5).

From Eq. (4),

$$\tan \delta = \frac{P_\delta}{Q_C} = \frac{\frac{V^2}{R_p}}{\omega C_p V^2} = \frac{1}{\omega C_p R_p} \tag{5}$$

The series equivalent circuit produces small $\tan \delta$ values, which are approximately equal, as can be seen in Eq. (6).

From Eq. (4),

$$\tan \delta = \frac{P_\delta}{Q_C} = \frac{R_s I^2}{\frac{I^2}{\omega C_s}} = \omega C_s R_s \tag{6}$$

2.2. Lightning impulse voltage

High voltage testing is necessary in order to evaluate the ability to withstand the effects of lightning impulses resulting in internal or external over voltages. Furthermore, the tests allow the breakdown mechanisms to be assessed. Generation of these high voltage impulses occurs through the rapid discharge of the charged capacitors via switching gaps and into the network comprising resistors and capacitors. In such scenarios, voltage multiplier circuits can frequently be employed [9, 10]. It is possible to measure the peak lightning values through the use of electronic circuits in combination with voltage dividers. The lightning property measurements are obtained through the use of an oscilloscope, so that the voltage time characteristics can be evaluated via voltage dividers or sometimes through analog digital converters [10].

This research study of lightning properties in the context of palm oil nanofluids indicates that voltage shapes are governed by the time parameters related to the front and tail. For the tests, the lightning impulse voltage used was in the form of a standard where the time of the wave front was $1.2 \mu\text{s} \pm 30\%$ and the time of the wave tail was $50 \mu\text{s} \pm 20\%$ on the basis of IEC 60897 [11].

2.3. Water content in palm oil

It is possible for water to dissolve in palm oil on a molecular level, although the solubility is minimal and dependent upon the temperature and the exact molecular composition. The solubility of water is greater in silicon oils and even more so in esters, especially phosphate esters, in which the level of solubility can be greater than 600 ppm. Absorption of the water by the palm oil may take place, or the process can result from the oxidation occurring in the palm oil [6]. This tendency of phosphate esters to absorb water is a challenge for users, since handling must be carried out with great care to avoid increasing the moisture content, which can rise to 0.1%, which will have adverse consequences in terms of the properties. This can be a problem for the dielectric strength and dissipation factor or $\tan \delta$.

This study therefore sought to manage the moisture levels in the samples. The measurements were taken using apparatus from Karl Fischer Titrators. The Karl Fischer reaction and ASTM D 1533 test were used to measure the water content in all of the different samples [12].

2.4. Palm oil and nanoparticles

This study conducted experiments using palm oil which had the brand name Morakot Palm Olein. Prior to use, this palm oil was prepared through the processes of RBD (refinement, bleaching, and deodorizing). It can be produced as a result of the fractionation of refined palm oil. In this process, two components are derived; the liquid part is known as olein, and the solid component is the stearin. In Thailand, RBD palm olein is widely employed for cooking purposes, both domestically and commercially. Other sources clearly list its various properties [13].

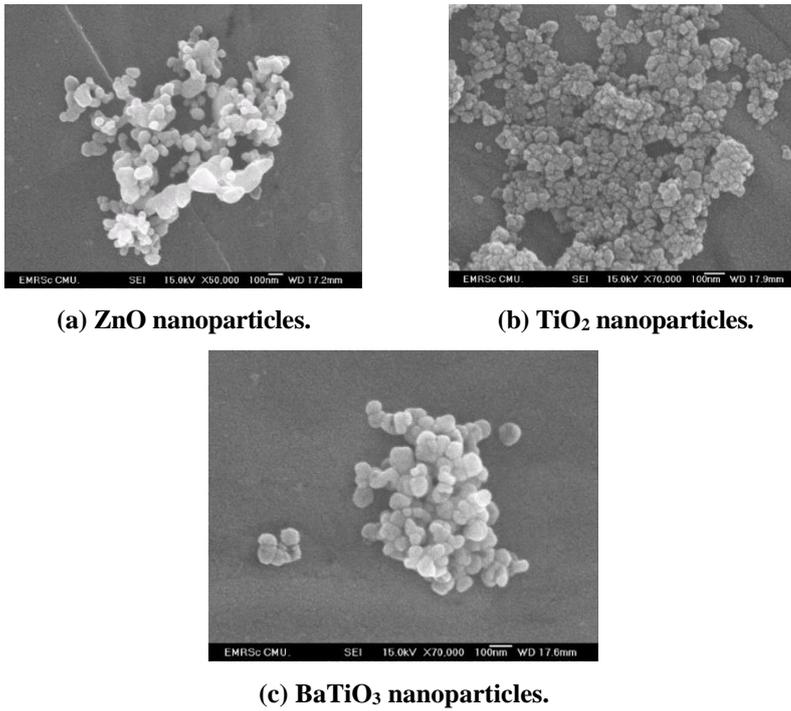
To produce palm oil-based nanofluids, the semi-conductive nanoparticles of ZnO and TiO₂. Alternatively, palm oil could be mixed with the dielectric nanoparticles of BaTiO₃. These particular ZnO, TiO₂, and BaTiO₃ nanoparticles had respective diameters of 30 nm, 40 nm, and 50 nm respectively [14]. SEM was employed to determine the structure and exact size of the dry nanopowder particles, and the results are presented in Fig. 3. It was clearly apparent that no particles had diameters greater than 50 nm.

3. Experimental Setup

3.1. Preparation of nanofluids

The nanofluid samples were produced via three different processes. First of all, the palm oil and the selected nanoparticles were combined using a 0.01% nanoparticle volume fraction. Additional samples were then produced using a 0.03% nanoparticle volume fraction. Next, the magnetic stirrer was dispersed nanofluids using for 30 minutes prior to a further two hours of ultrasonic dispersal to ensure nanofluid homogeneity. Magnetic stirring is particularly effective in ensuring that the nanoparticles are evenly distributed in the base oil, while energy levels are maintained below the point which could cause the nanoparticle agglomerations to

loss. To achieve such nanofluids samples, ultrasonic dispersal is necessary [15, 16]. Figure 4 shows the process of preparing nanofluids.



(a) ZnO nanoparticles.

(b) TiO₂ nanoparticles.

(c) BaTiO₃ nanoparticles.

Fig. 3. SEM analysis of the sample of the nanoparticle surface.

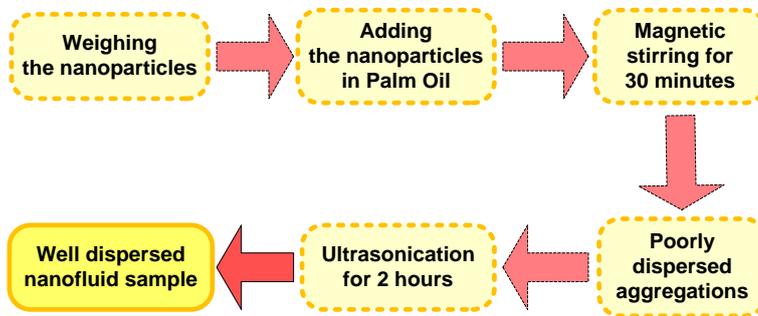


Fig. 4. Combining palm oil with nanoparticles.

3.2. Instrumentation used to measure the properties of $\tan \delta$

Measuring bridges are specifically designed to draw magnitude and phase comparisons between currents which flow through the test object and then through an almost totally lossless standard capacitor. Around 100 years ago, a high-voltage bridge designed to measure the dielectric dissipation factor was created by

Schering. This original design remains in use, with certain modifications [17]. There will be a measurable response in the $\tan \delta$ values in terms of insulation capabilities and palm oil ageing when using SOKEN DAC-IM-D6 as the $\tan \delta$ tester model. The value is therefore a critical guide to transformer operation. This research assessed $\tan \delta$ in pure or unmodified palm oil and nanofluids based on palm oil at each of the following temperatures: 35 °C to 90 °C. Measurement processes followed the IEC 60247 standards [18]. Figures 5 and 6 presents the set-up of the testing system.

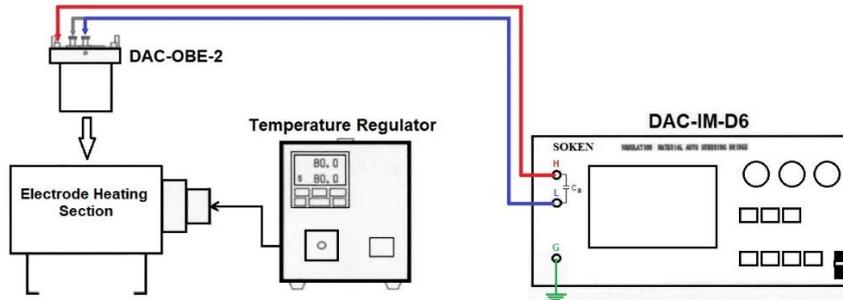


Fig. 5. The test circuit connections for $\tan \delta$ measurement.

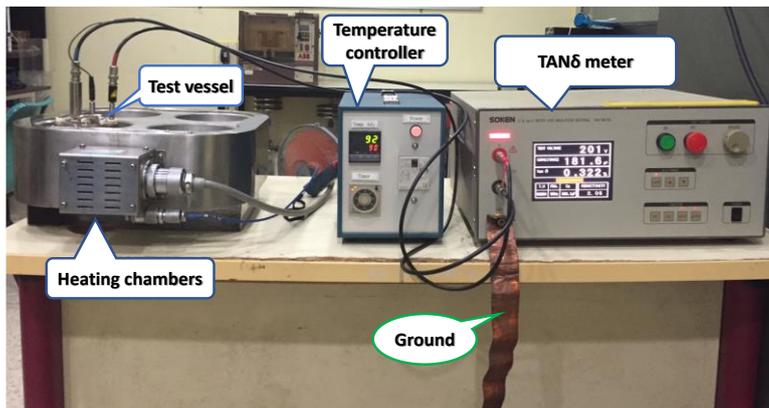


Fig. 6. The test circuit set-up used for the measurement of $\tan \delta$ characteristics.

3.3. Evaluation of lightning properties

Testing of the lightning properties was conducted using a needle-sphere electrode system in line with the guidelines of IEC 60897 [11]. The test circuit diagram is shown in Fig. 7, and the experimental testing set up can be seen in Fig. 8. Gap spacing for the experimental configuration was set to be 15 mm for the needle-sphere electrodes: a tungsten needle whose tip radius was 40 μm served as the high voltage electrode while a brass sphere of 13 mm in diameter served as the grounded electrode. On the basis of IEC 60897, a rising voltage technique was used. In each test, the voltage level at the start was gradually and constantly increased until the eventual breakdown took place. The voltage increases were performed in incremental steps of 5 kV. A time interval of 60 seconds was established between

each breakdown. With no changes made to the fluid samples or the electrodes, it was possible to complete six breakdowns for each of the samples and for the results to be recorded.

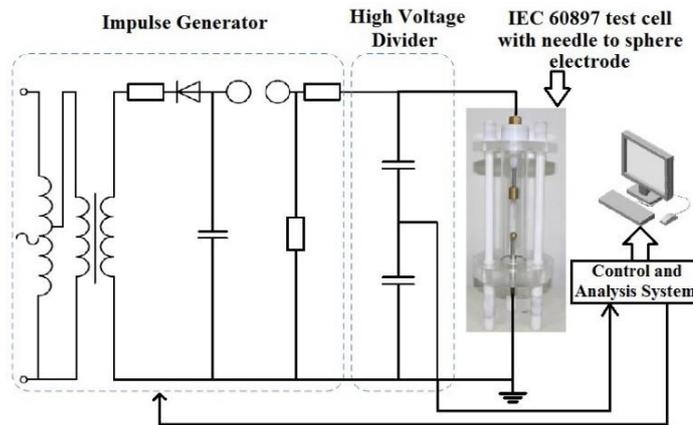


Fig. 7. The test circuit connections for lightning properties.

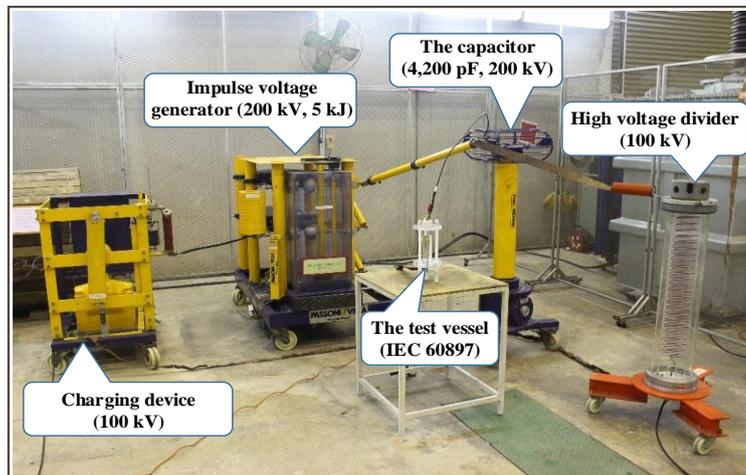


Fig. 8. The test circuit set-up used for the measurement of lightning properties.

4. Results and Analysis

4.1 Test results for water content

The presence of nanoparticles has a clear effect upon the water levels. Water content test results are presented in Fig. 9 and show findings for both pure or unmodified palm oil and nanofluids based on palm oil. These tests were carried out prior to commencing experimentation. These results confirmed the unmodified palm oil contained higher levels of moisture than the samples of palm oil with nanoparticles.

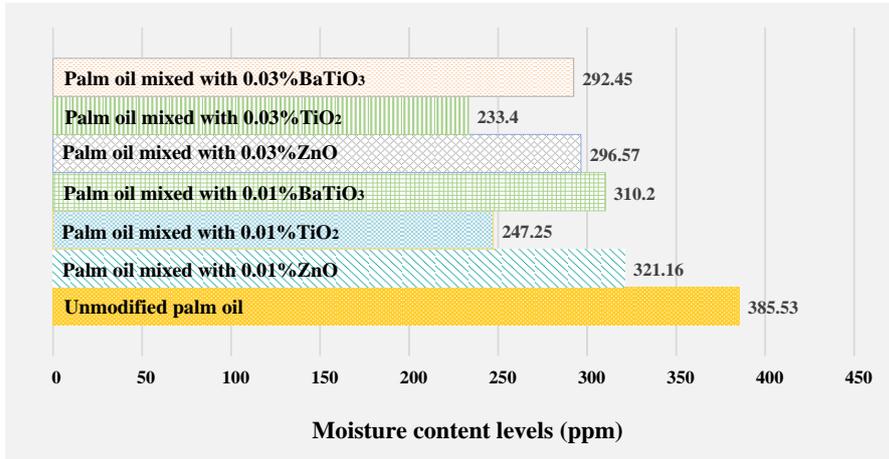


Fig. 9. Results for water content measurements.

The initial differences may contribute to the later $\tan \delta$ values and lightning properties recorded in palm oils, as any subsequent changes may depend on whether or not the palm oil contains nanoparticles. One particular effect is that $\tan \delta$ and the lightning properties are influenced by electron traps which can occur when electron transfer takes place. It is also possible that water content itself can affect $\tan \delta$, which is also governed by temperature changes inside the electrode (DAC-OBE-2) and the lightning properties by the needle to sphere electrode within test cell.

4.2 Test results for properties of $\tan \delta$

Figure 10 shows the outcomes for $\tan \delta$ in unmodified palm oil and in palm oil-based nanofluids which use different temperatures and different nanoparticle concentrations. From these results, it is possible to make comparisons. As the temperature rises incrementally from 35 °C to 90 °C, the $\tan \delta$ also rises for all samples. In all cases, there was an increase in $\tan \delta$ as temperatures increased. Moreover, the exact influence of the various nanofluids upon $\tan \delta$ in terms of the notion that certain quantities of certain nanoparticles might increase the dielectric qualities of palm oil. It is clear that for all nanofluids, the $\tan \delta$ values are higher than those of unmodified palm oil. Furthermore, the $\tan \delta$ values are also affected by nanofluid concentrations. As temperature rises, the $\tan \delta$ values increase, and when the concentrations are higher, this also leads to increased $\tan \delta$ values, especially when the concentration of added ZnO were 0.01% and 0.03%.

Besides, the dissipation factor $\tan \delta$ is a material parameter that is determined by the polarization and conductivity losses, according to Eq. (4), Fig. 1. Furthermore, moisture causes a significant rise in losses due to quickly polarized water molecules as well as an increase in conductivity. This is particularly important for moisture-sensitive materials like palm oil and filled synthetics [8]. The addition of ZnO, TiO₂, and BaTiO₃ to palm oil resulted in a decrease in moisture content in this study.

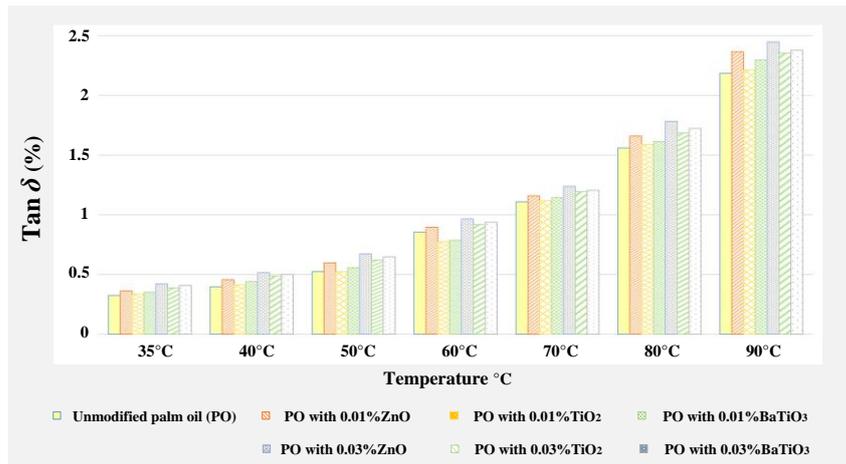


Fig. 10. Comparison of $\tan \delta$ results for unmodified palm oil and palm oil based nanofluids.

The palm oil-based nanofluids can serve as a heat transfer medium while still being electrically insulating. In enhancing the heat transfer of transformers, thermal conductivity properties are considered a critical parameter [19]. Moreover, its thermal conductivity properties and suspended nanomaterials are expected to improve heat transfer and thermal conductivity efficiency by reducing power loss and evaporation losses. The thermal conductivity of palm oil is affected by the suspension of ZnO, TiO₂, and BaTiO₃ nanoparticles, and suspended nanoparticles are supposed to improve thermal conductivity and heat transfer efficiency.

4.3 Lightning properties testing results

The breakdown voltage strength for the lightning impulses was measured in order to establish the related properties of the palm oil. The mean values for the breakdown records were used as a means of estimating the unmodified palm oil quality, and hence the quality of the nanofluids derived from the palm oil. However, the design of the actual transformers is not based on mean values; instead, the minimum voltage which can be withstood is the value which is considered to be more important. Experimental studies have shown that variables such as nanoparticle types and nanoparticle volume concentration influence impulse breakdown strength. The mechanism by which nanoparticles influence oil breakdown properties is yet to be thoroughly elucidated. The traditional theory of liquid dielectric dissolution fails to account for the improved insulating properties of palm oil-based nanofluids [20]. The scattered conductive nanoparticles absorb the electrons and reduce their mobility and energy transfer, according to Hwang et al. [21]. This was the primary cause of improved dielectric efficiency and lower positive streamer velocity in the prepared nanofluids. Suspended nanoparticles capture high mobility electrons created by electric field ionization, which are then converted to slow negative charge carriers. The charge relaxation time constant (τ_r) theory was used to further explain the main electrodynamic mechanism for palm oil and nanoparticle materials. Equation (7) gives the relaxing time constant [21, 22].

$$\tau_r = \frac{2\varepsilon_1 + \varepsilon_2}{2\kappa_1 + \kappa_2} \quad (7)$$

where ε_1 and ε_2 represent the permittivity of palm oil and nanoparticle materials respectively; κ_1 and κ_2 represent the electrical conductivities of palm oil and nanoparticle materials respectively. Furthermore, a low relaxation time constant indicates rapid electron absorption on the nanoparticles' surface.

A summary of the mean lightning impulse breakdown voltage strengths can be seen in Fig. 11, while Tables 1 and 2, include presents the percentage increments for the nanofluid lightning strengths. In all cases, the addition of nanoparticles leads to increases in both positive and negative lightning properties when compared to the results for unmodified palm oil. Further analysis is necessary to determine the exact types and volume fractions of nanoparticles which will deliver optimal results for lightning properties. To analyse the effect of nanoparticles added in the palm oil on the dielectric strength's property. The breakdown strength of the palm oil-based nanofluid increased as a result of the nanoparticles ZnO, TiO₂, and BaTiO₃ obstructing streamer growth and decreasing streamer propagation [16]. The effect of the concentrations of ZnO, TiO₂, and BaTiO₃ nanoparticles can be explained as follows. The spacing between nanoparticles in palm oil decreased, as the concentration of nanoparticles mixed in palm oil increased. Therefore, there was a possibility that electrons could be accelerated and ionized, allowing them to ionize oil molecules. This explains why, at a concentration of 0.01%, the breakdown intensity improved marginally. The spacing between particles grew smaller as the concentration of ZnO, TiO₂, and BaTiO₃ nanoparticles increased, and electrons were caught quickly in the oil gap. Meanwhile, at 0.03% ZnO, TiO₂, and BaTiO₃ concentrations, this resulted in a significant increase in breakdown strength.

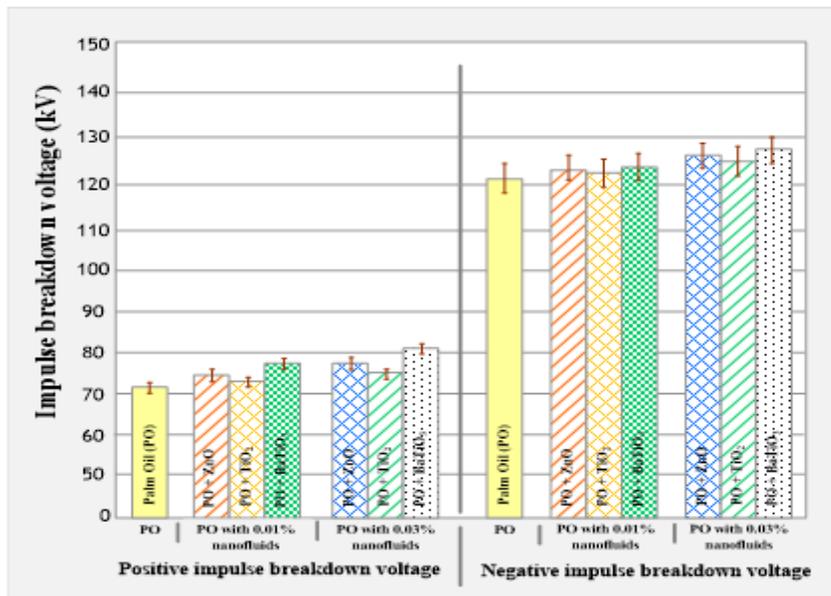


Fig. 11. Comparison of the average lightning breakdown voltages of unmodified palm oil and palm oil based nanofluids.

Table 1. Results of positive lightning strength of unmodified palm oil and palm oil mixed with 0.01% and 0.03% ZnO, TiO₂, and BaTiO₃.

Positive lightning strength (kV)				% Increase		
Palm oil pure	Palm oil mixed with 0.01% nanofluids			ZnO	TiO ₂	BaTiO ₃
	ZnO	TiO ₂	BaTiO ₃			
71.03	75.73	74.53	78.55	6.61	4.92	10.58
Palm oil pure	Palm oil mixed with 0.03% nanofluids			ZnO	TiO ₂	BaTiO ₃
71.03	78.51	76.86	80.48	10.53	8.20	13.30

Table 2. Results of negative lightning strength of unmodified palm oil and palm oil mixed with 0.01% and 0.03% ZnO, TiO₂, and BaTiO₃.

Negative lightning strength (kV)				% Increase		
Palm oil pure	Palm oil mixed with 0.01% nanofluids			ZnO	TiO ₂	BaTiO ₃
	ZnO	TiO ₂	BaTiO ₃			
121.50	122.90	122.35	123.53	1.15	0.70	1.67
Palm oil pure	Palm oil mixed with 0.03% nanofluids			ZnO	TiO ₂	BaTiO ₃
121.50	126.01	124.70	127.79	3.71	2.63	5.17

In addition, the evident in Tables 3 and 4 that the nanofluids based on palm oil has higher the lightning strength in comparison with pure palm oil in all samples. The nanofluids based on palm oil indicated overall the best performance of the lightning properties results, demonstrating higher mean the lightning strength and lower standard deviation. A high standard deviation indicates that the values on average, in Table 3 (group 0.01%), are further from the mean, especially in the case of palm oil mixed with 0.01% BaTiO₃. Besides, a small standard deviation in Table 4 (group 0.03%) indicates that the values in a statistical data set on test results, on average, similar to the data set's mean.

Table 3. Results of mean, standard deviation, lowest and maximum of positive the lightning strength (LS) values.

Dielectric liquid	Lowest LS, kV	Maximum LS, kV	Mean LS, kV	Std. deviation, kV	
Palm oil (PO)	68.40	73.63	71.03	2.01	
Group 0.01%	PO+ZnO	72.02	79.26	75.73	2.43
	PO+TiO ₂	72.58	78.72	74.53	2.31
	PO+BaTiO ₃	75.46	83.18	78.55	3.45
Group 0.03%	PO+ZnO	76.76	80.28	78.51	1.23
	PO+TiO ₂	74.81	78.72	76.86	1.44
	PO+BaTiO ₃	79.28	82.63	80.48	1.19

Table 4. Results of mean, standard deviation, lowest and maximum of negative the lightning strength (LS) values.

Dielectric liquid	Lowest LS, kV	Maximum LS, kV	Mean LS, kV	Std. deviation, kV	
Palm oil (PO)	121.16	122.05	121.50	0.29	
Group 0.01%	PO+ZnO	121.44	124.22	122.90	1.18
	PO+TiO ₂	121.44	123.50	122.35	0.78
	PO+BaTiO ₃	122.61	125.63	123.53	1.17
Group 0.03%	PO+ZnO	125.63	126.74	126.06	0.50
	PO+TiO ₂	123.70	125.68	124.70	0.79
	PO+BaTiO ₃	126.85	128.91	127.79	0.79

5. Conclusions

This study provided an evaluation of the effects of nanoparticles in the context of $\tan \delta$, or the dielectric dissipation factor at a range of different temperatures and the lightning properties was performed at the room temperature, for nanofluids based on palm oil. The results showed that nanofluid $\tan \delta$ properties were significantly influenced in the case of all of the different nanoparticle types: ZnO, TiO₂, and BaTiO₃. The dielectric loss tangent is shown by the sample $\tan \delta$ values, and the increases were significant for nanofluids based on palm oil when compared to unmodified palm oil. Besides, data analysis revealed that the lightning properties of palm oil nanofluids depend significantly upon the specific type of nanoparticles introduced to the palm oil. There were three nanoparticle types used: BaTiO₃, ZnO, and TiO₂. When BaTiO₃ is added to the palm oil, the result is enhanced lightning strength with both positive and negative polarity. In the case of ZnO or TiO₂ nanoparticles, there is only a small change recorded in the lightning strength. This confirmed that the nanoparticles do indeed have an influence upon $\tan \delta$ and lightning properties in palm oil-based nanofluids. Therefore, the nanofluids based on palm oil should undergo further investigation to assess the potential for further enhancements. Moreover, these results suggested that further investigation would be worthwhile to better understand the effects of the nanoparticles in palm oil upon $\tan \delta$ and the lightning strength in order to make additional improvements.

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Nomenclatures

C	Capacitance
C_p	Parallel equivalent capacitance
C_s	Series equivalent capacitance
I	Current, complex r.m.s. phasor
I_C	Current, comprising capacitive elements
I_δ	Current, comprising resistive elements
P_δ	Dielectric power loss
Q_C	Capacitive reactive power in a dielectric
R_p	Parallel equivalent resistance
R_s	Series equivalent resistance
V	Voltage, complex r.m.s. phasor

Greek Symbols

κ	Electrical conductivity
φ	Phase angle
ω	Angular frequency

ε	Permittivity
τ_r	Charge relaxation time constant
δ	Loss angle
Abbreviations	
ASTM	American Society for Testing and Materials
IEC	International Electrotechnical Commission
SEM	Scanning Electron Microscopy

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